The Synthesis of Difluoro[1]benzothieno[2,3-c]quinolines and Their N-Methyl Quaternary Salts

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A series of difluoro[1]benzothieno[2,3-c]quinolines has been prepared by photocyclization of the appropriate carboxamides. The lactams obtained were converted into the corresponding chloro derivatives which were catalytically dechlorinated into the difluoro[1] benzothieno[2,3-c]quinolines. The latter compounds were transformed into the N-methyl quaternary salts.

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We have been interested in the synthesis of polycyclic nitrogen and sulfur containing heterocycles via photocyclization of the appropriate anilides [2-6]. This led us to the discovery of the antileukemic activity against L-1210 of 5-methyl[1]benzothieno[2,3-c]quinolinium iodide 1 [4,7] and its 4,10-dimethoxy-substituted derivative 2 [2,7] as sulfurcontaining analogs of the naturally occurring antileukemic alkaloids fagaronine 3 [8] and nitidine 4 [9]. This

finding prompted us to report the synthesis of a series of difluoro[1]benzothieno[2,3-c]quinolines 32-37 and their N-methyl quaternary salts 38-43 for biological evaluation since the introduction of a fluorine atom into an organic compound has frequently produced a change in biological activity due to the stability of the C-F bond, the strong electronegativity of the fluorine atom, and the size of the fluorine atom similar to that of the size of the hydrogen atom.

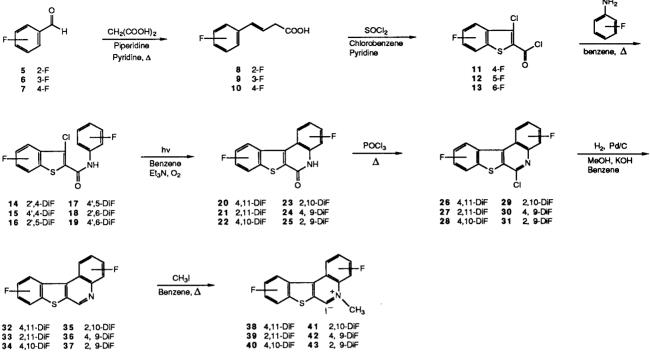
Thus o-fluorobenzaldehyde 5, m-fluorobenzaldehyde 6 or p-fluorobenzaldehyde 7 were allowed to react with malonic acid under Knoevenagel condensation conditions [10] to give 2-, 3-, or 4-fluorocinnamic acids 8 (73%), 9 (80%), or 10 (73%), respectively. Preparation of the fluorine-substituted 3-chlorobenzo[b]thiophene-2-carbonyl chlorides 11, 12, and 13 was carried out in a similar manner to that described by Wright and Brabander [11] starting with fluorocinnamic acids 8, 9 and 10. When 3-chloro-4-fluorobenzo[b]thiophene-2-carbonyl chloride 11 was

treated with o- or p-fluoroaniline, the corresponding carboxamides 14 (92%) and 15 (92%) were isolated. Likewise, treatment of 12 or 13 with o- or p-fluoroaniline in benzene solution afforded carboxamides 16 (88%), 17 (91%), 18 (83%), and 19 (85%), respectively.

Oxidative photocyclization of the above carboxamides was accomplished by irradiating a mixture of equimolar amounts of the carboxamide and triethylamine with a 450 watt medium pressure mercury vapor lamp using benzene as the solvent to yield the corresponding difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-ones 20 (93%), 21 (98%), 22 (86%), 23 (70%), 24 (52%), and 25 (48%), respectively. The lactams were chlorinated in refluxing phosphorus oxychloride to provide difluoro-6-chloro[1]benzothieno-[2,3-c]quinolines 26 (37%), 27 (83%), 28 (84%), 29 (67%), 30 (74%), and 31 (60%).

Catalytic dechlorination of 26-31 with hydrogen and 10% Pd-C as the catalyst in 1:1 benzene-methanol solution in the presence of potassium hydroxide resulted in difluoro[1]benzothieno[2,3-c]quinolines 32 (78%), 33 (74%), 34 (82%), 35 (82%), 36 (83%), and 37 (89%). When 32-37 were allowed to react with iodomethane in benzene in a sealed tube at 100-110°, the difluoro-5-methyl[1]benzothieno[2.3-clauinolinium iodides 38 (68%), 39 (71%), 40 (24%), 41 (74%), 42 (55%), and 43 (73%) were obtained. Interestingly, the ¹H nmr spectra of compounds 38, 40 and 42 exhibited a doublet centered at ca. δ 4.84 ppm with J =10 Hz for the N-methyl group. The relative magnitudes of the 1H-19F coupling constants of fluorinated aromatic compounds are remarkably constant [12]. We observe (Table 1) that the heteronuclear coupling constants between the H1 resonance and the fluorine at the 11-position are larger in every case than the corresponding vicinal coupling between H1 and a fluorine in the 2-position. In light of this, it is difficult to believe that a scalar coupling through six bonds would be larger than the corresponding coupling through three bonds when the latter is contained wholly within the same aromatic ring. Given this, coupled with the fact that H1 and the 11-19F substituent are in proximity in a bay-region, it seems logical to conclude that these substituents may be coupled through space rather than through bonds as would normally be the case.





These quaternary salts will be submitted for biological screening and these data will be reported elsewhere.

Compound Number	$J_{1,F_{11}}$ in Hz	J_{1,F_2} in Hz
21	11.7	9.0
27	12.0	9.3
33	12.2	10.5

EXPERIMENTAL

Melting points were obtained on a Thomas-Hoover melting point apparatus and are uncorrected. The ir spectra were recorded on a Beckman FT 1100 spectrometer as potassium bromide pellets and frequencies are expressed in cm⁻¹. The ¹H nmr spectra were obtained on a JEOL FX-90Q spectrometer in the solvent indicated with TMS as the internal standard. Chemical shifts are reported in ppm (δ) and J values in Hz. Elemental analyses were performed by M-H-W Laboratories, Phoenix, Arizona.

2-Fluorocinnamic Acid (8).

A mixture of 100.0 g (0.81 mole) of 2-fluorobenzaldehyde (5), 109.1 g (1.05 moles) of malonic acid, 15 ml of piperidine, and 1000 ml of pyridine was heated at 100° for 20 hours. After cooling to room temperature, the solution was poured into ca. 3000 ml of ice water and acidified to pH 3-4 with ca. 900 ml of concentrated hydrochloric acid. The precipitate was collected by filtration and washed with water and then recrystallized from 95% ethanol with charcoal as a decolorizing agent to give 98.5 g (0.59 mole, 73%) of 8, mp 178-179°; lit 175° [13,14], 177-178° [15]; ¹H

nmr (DMSO-d₆): δ 6.52 (d, J = 16.1 Hz, 1H, CH-CO₂H), 6.97-7.66 (m, 4H, ArH), 7.80 (d, J = 16.1 Hz, 1H, CH=CHCO₂H), 10.29 (br s, 1H, CO₂H).

3-Fluorocinnamic Acid (9).

This compound was prepared from 3-fluorobenzaldehyde (6) (50.8 g, 0.41 mole) and malonic acid (54.0 g, 0.52 mole) in piperidine (7.5 ml) and pyridine (500 ml) in a manner similar to that described for the synthesis of **8**, thus 55.0 g (0.33 mole, 80%) of **9** was obtained, mp 166-167°, lit [16-17] 166.5°; ¹H nmr (DMSO-d₆): δ 6.42 (d, J = 16.2 Hz, 1H, CH-CO₂H), 6.95-7.50 (m, 4H, ArH), 7.61 (d, J = 16.2 Hz, 1H, CH = CHCO₂H), 9.05 (br s, 1H, CO₂H).

4-Fluorocinnamic Acid (10).

This compound was prepared from 4-fluorobenzaldehyde (7) (101.4 g, 0.82 mole) and malonic acid (108.4 g, 1.04 moles) in piperidine (15 ml) and pyridine (1000 ml) in a manner similar to that described for the synthesis of $\bf 8$; 98.9 g (0.60 mole, 73%) of $\bf 10$ was obtained, mp 206-208°, lit 203° [18], 206-207° [19]; 'H nmr (DMSO-d₆): δ 6.26 (d, J = 15.9 Hz, 1H, CHCO₂H), 6.89-7.60 (m, 5H, ArH and CH=CHCO₂H), 9.05 (br s, 1H, CO₂H).

3-Chloro-4-fluorobenzo[b]thiophene-2-carbonyl Chloride (11) [11].

To a suspension of 94.0 g (0.57 mole) of 8 in 500 ml of chlorobenzene containing 4.26 ml of pyridine in an ice bath was added dropwise 200 ml (2.74 moles) of thionyl chloride. The mixture was gradually warmed to room temperature and then heated under reflux for 80 hours. The solvent and the excess thionyl chloride were removed by distillation under reduced pressure and the residue was triturated with hexanes (350 ml). The solid was collected by filtration and recrystallized from hexanes to give 32.0 g (0.13 mole, 23%) of 11 as yellow crystals, mp 126.5-128°; ¹H nmr (deuteriochloroform): δ 7.03-7.26 (m, 1H, H-6), 7.42-7.67 (m, 2H, H-5 and H-7).

3-Chloro-5-fluorobenzo[b]thiophene-2-carbonyl Chloride (12) [11].

This compound was prepared from **9** (53.0 g, 0.32 mole), thionyl chloride (110 ml, 1.51 moles) in chlorobenzene (300 ml) and pyridine (2.4 ml) in a manner similar to that described for the synthesis of **11**; 21.0 g (84 mmoles, 26%) of **12** was obtained as yellow crystals, mp 104-106° lit [11] 108-110°; 'H nmr (deuteriochloroform): δ 7.26-7.48 (m, 1H, H-6), 7.63 (dd, $J_{4,F} = 8.8$ Hz, $J_{4,6} = 2.4$ Hz, 1H, H-4), 7.80 (dd, $J_{6,7} = 8.8$ Hz, $J_{F,7} = 4.6$ Hz, 1H, H-7).

3-Chloro-6-fluorobenzo[b]thiophene-2-carbonyl Chloride (13) [11].

This compound was prepared from 10 (98.9 g, 0.60 mole), thionyl chloride (215 ml, 2.95 moles) in chlorobenzene (500 ml) and pyridine (5 ml) in a manner similar to that described for the synthesis of 11, thus 49.2 g (0.20 mole, 33%) of 13 was obtained as yellow needles, mp 106-108° lit [11] 109-111°; ¹H nmr (deuteriochloroform): δ 7.16-7.39 (m, 1H, H-5), 7.50 (dd, $J_{F,7} = 8.1$ Hz, $J_{5,7} = 2.4$ Hz, 1H, H-7), 7.96 (dd, $J_{4,5} = 9.0$ Hz, $J_{4,F} = 5.1$ Hz, 1H, H-4).

3-Chloro-4-fluoro-N(2'-fluorophenyl)benzo[b]thiophene-2-carboxamide (14).

A mixture of 6.0 g (24 mmoles) of 11, 2.68 g (24 mmoles) of ofluoroaniline in 85 ml of benzene was heated under reflux for two hours. After cooling to room temperature the solvent was removed under reduced pressure and the residue was recrystallized from 95% ethanol to give 7.09 g (22 mmoles, 92%) of 14 as offwhite needles, mp 150-152°; ir (potassium bromide): 3394 (NH stretching), 1669 (C=0 stretching); ¹H nmr (DMSO-d₆): δ 7.14-7.67 (m, 5H, ArH), 7.76-7.97 (m, 2H, ArH), 9.92 (br s, 1H, NH).

Anal. Calcd. for $C_{15}H_aClF_2NOS$: C, 55.65; H, 2.49; N, 4.33; S, 9.90. Found: C, 55.68; H, 2.46; N, 4.33; S, 9.89.

3-Chloro-4-fluoro-N-(4'-fluorophenyl)benzo[b]thiophene-2-carboxamide (15).

This compound was prepared from 11 (6.0 g, 24 mmoles) and p-fluoroaniline (2.68 g, 24 mmoles) in benzene (85 ml) in a manner similar to that described for the synthesis of 14, and 7.05 g (22 mmoles, 92%) of 15 was obtained from 95% ethanol as off-white needles, mp 168-170°; ir (potassium bromide): 3302 (NH stretching), 1640 (C=O stretching); ¹H nmr (DMSO-d₆): δ 7.13-7.85 (m, 6H, ArH), 7.98 (dd, $J_{6,7} = 8.0$ Hz, $J_{5,7} = 1.3$ Hz, 1H, H-7), 10.67 (br s, 1H, NH).

Anal. Calcd. for C₁₅H₈ClF₂NOS: C, 55.65; H, 2.49; N, 4.33; S, 9.90. Found: C, 55.49; H, 2.50; N, 4.29; S, 10.00.

3-Chloro-5-fluoro-N-(2'-fluorophenyl)benzo[b]thiophene-2-carboxamide (16).

This compound was prepared from 12 (4.0 g, 16 mmoles) and o-fluoroaniline (1.78 g, 16 mmoles) in benzene (60 ml) in a manner similar to that described for the synthesis of 14, therefore 4.57 g (14 mmoles, 88%) of 16 was obtained from cyclohexane as colorless crystals, mp 162-163°; ir (potassium bromide): 3394 (NH stretching), 1671 (C=O stretching); 1 H nmr (DMSO-d₆): 100° δ 7.18-8.00 (m, 6H, ArH), 8.14 (dd, $J_{6,7} = 9.2$ Hz, $J_{F,7} = 4.8$ Hz, 1H, H-7), 9.87 (br s, 1H, NH).

Anal. Calcd. for C₁₅H_aClF₂NOS: C, 55.65; H, 2.49; N, 4.33; S, 9.90. Found: C, 55.88; H, 2.74; N, 4.35; S, 10.01.

3-Chloro-5-fluoro-N-(4'-fluorophenyl)benzo[b]thiophene-2-carboxamide (17).

This compound was prepared from 12 (4.0 g, 16 mmoles) and p-fluoroaniline (1.78 g, 16 mmoles) in benzene (60 ml) in a manner similar to that described for the synthesis of 14; 4.73 g (14.6 mmoles, 91%) of 17 was obtained from benzene as fine colorless needles, mp 180-182°; ir (potassium bromide): 3296 (NH stretching), 1640 (C = 0 stretching); ¹H nmr (DMSO-d₆): δ 7.11-7.84 (m, 6H, ArH), 8.19 (dd, $J_{6,7} = 9.0$ Hz, $J_{F,7} = 4.8$ Hz, 1H, H-7), 10.60 (br s, 1H, NH).

Anal. Calcd. for C₁₈H₈ClF₂NOS: C, 55.65; H, 2.49; N, 4.33; S, 9.90. Found: C, 55.50; H, 2.70; N, 4.16; S, 9.79.

3-Chloro-6-fluoro-N-(2'-fluorophenyl)benzo[b]thiophene-2-carboxamide (18).

This compound was prepared from 13 (5.0 g, 20 mmoles) and o-fluoroaniline (2.23 g, 20 mmoles) in benzene (72 ml) in a manner similar to that described for the synthesis of 14, and 5.34 g (16.5 mmoles, 83%) of 18 was obtained from 95% ethanol as off-white needles, mp 170-172°; ir (potassium bromide): 3386 (NH stretching), 1661 (C=O stretching); ¹H nmr (DMSO-d₆): δ 7.20-7.60 (m, 4H, ArH), 7.74-7.99 (m, 2H, ArH), 8.09 (dd, $J_{F,7}$ = 9.0 Hz, $J_{5,7}$ = 2.2 Hz, 1H, H-7), 10.13 (br s, 1H, NH).

Anal. Caled. for C₁₅H₈ClF₂NOS: C, 55.65; H, 2.49; N, 4.33; S, 9.90. Found: C, 55.74; H, 2.65; N, 4.35; S, 9.83.

3-Chloro-6-fluoro-N-(4'-fluorophenyl)benzo[b]thiophene-2-carboxamide (19).

This compound was prepared from 13 (5.0 g, 20 mmoles) and p-fluoroaniline (2.23 g) in benzene (72 ml) in a manner similar to that described for the synthesis of 14, thus 5.52 g (17 mmoles, 85%) of 19 was obtained from benzene as off-white needles, mp 178-181°; ir (potassium bromide): 3317 (NH stretching), 1643 (C=0 stretching); 'H nmr (DMSO-d₆): δ 7.12-7.99 (m, 6H, ArH), 8.10 (dd, $J_{F,7} = 9.0$ Hz, $J_{5,7} = 2.2$ Hz, 1H, H-7), 10.55 (br s, 1H, NH).

Anal. Caled. for C₁₅H₈ClF₂NOS: C, 55.65; H, 2.49; N, 4.33; S, 9.90. Found: C, 55.68; H, 2.54; N, 4.33; S, 9.86.

4,11-Difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-one (20).

A solution of 0.5 g (1.54 mmoles) of 14 in 500 ml of benzene containing 0.15 g of triethylamine was irradiated with a 450 watt Hanovia medium pressure mercury vapor lamp for 4 hours. A slow stream of air was passed through the solution during the course of the reaction. The solid was collected by filtration and washed with water to give 0.41 g (1.43 mmoles, 93%) of 20 as an off-white powder, mp >280°; ir (potassium bromdie): 3142 (NH stretching), 1666 (C=O stretching): 1 H nmr (DMSO-d₆): 140° δ 7.25-7.78 (m, 4H, ArH), 8.01 (dd, J_{8,9} = 7.9 Hz, J_{8,10} = 1.0 Hz, 1H, H-8), 8.47 (m, 1H, H-1), 11.50 (br s, 1H, NH). This compound was used without further purification in the next reaction because of low solubility.

2,11-Difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-one (21).

This compound was obtained in 98% yield from 15 in a manner similar to that described for the synthesis of 20, mp $>280^\circ$; ir (potassium bromide): 3137 (NH stretching), 1669 (C = O stretching); ¹H nmr (DMSO-d₆): 150° δ 7.28-7.77 (m, 4H, ArH), 8.01 (dd, $J_{8,9}=8.0$ Hz, $J_{8,10}=0.9$ Hz, 1H, H-8), 8.39 (ddd, $J_{1,F11}=11.7$ Hz, $J_{1,F2}=9.0$ Hz, $J_{1,3}=2.7$ Hz, 1H, H-1), 11.84 (br s, 1H, NH). This compound was used without further purification in the next

reaction due to its low solubility.

4,10-Difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-one (22).

This compound was obtained in 86% yield from 16 in a manner similar to that described for the synthesis of 20, mp > 280°; ir (potassium bromide): 3145 (NH stretching), 1664 (C = 0 stretching); ¹H nmr (DMSO-d₆): 150° δ 7.30-7.61 (m, 3H, ArH), 8.22 (dd, $J_{8.9} = 9.0 \text{ Hz}, J_{8.F} = 5.3 \text{ Hz}, 1\text{H}, \text{H-8}, 8.34-8.58 (m, 2\text{H}, \text{H-1}) and$ H-11), 11.81 (br s, 1H, NH). This compound was used without further purification in the next reaction due to its low solubility.

2.10-Difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-one (23).

This compound was obtained in 70% yield from 17 in a manner similar to that described for the synthesis of 20, mp > 280°; ir (potassium bromide): 3137 (NH stretching), 1653 (C = O stretching): ¹H nmr (DMSO-d₆): 150° δ 7.25-7.67 (m, 3H, ArH), 8.20 (dd, $J_{8.9} = 8.7 \text{ Hz}, J_{8.F} = 5.4 \text{ Hz}, 1\text{H}, H-8), 8.28 \text{ (dd}, J_{1.F} = 10.3 \text{ Hz},$ $J_{1.3} = 2.6 \text{ Hz}, 1H, H-1), 8.50 \text{ (dd}, J_{F,11} = 10.8 \text{ Hz}, J_{9,11} = 2.3 \text{ Hz},$ 1H, H-11), 11.85 (br s, 1H, NH). This compound was used without further purification in the next reaction due to its low solubility. 4.9-Difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-one (24).

This compound was obtained in 52% yield from 18 in a manner similar to that described for the synthesis of 20, mp > 280°; ir (potassium bromide): 3145 (NH stretching), 1643 (C = 0 stretching); ¹H nmr (DMSO-d₆): 120° δ 7.28-7.58 (m, 3H, ArH), 8.05 (dd, $J_{8.F} = 9.0 \text{ Hz}, J_{8.10} = 2.6 \text{ Hz}, 1\text{H}, H-8), 8.42 \text{ (m, 1H, H-1)}, 8.80 \text{ (dd, }$ $J_{10,11} = 9.3 \text{ Hz}, J_{F,11} = 5.1 \text{ Hz}, 1\text{H}, H-11), 11.77 \text{ (br s, 1H, NH)}.$ This compound was used without further purification in the next reaction due to its low solubility.

2.9-Difluoro[1]benzothieno[2,3-c]quinolin-6(5H)-one (25).

This compound was obtained in 48% yield from 19 in a manner similar to that described for the synthesis of 20, mp >280°; ir (potassium bromide): 3137 (NH stretching), 1653 (C = O stretching); ¹H nmr (DMSO-d₆): 150° δ 7.25-7.67 (m, 3H, ArH), 8.01 (dd, $J_{8,F} = 8.9 \text{ Hz}, J_{8,10} = 2.5 \text{ Hz}, 1H, H-8), 8.30 (dd, J_{1,F} = 10.4 \text{ Hz},$ $J_{1.3} = 2.5 \text{ Hz}, 1H, H-1), 8.78 (dd, J_{10,11} = 9.1 \text{ Hz}, J_{F,11} = 5.0 \text{ Hz},$ 1H, H-11), 11.68 (br s, 1H, NH). This compound was used without further purification in the next reaction due to its low solubility.

6-Chloro-4,11-difluoro[1]benzothieno[2,3-c]quinoline (26).

A mixture of 2.03 g (7.07 mmoles) of lactam 20 and 45 ml of phosphorus oxychloride was heated at 100-110° for four hours. After cooling in an ice bath, the mixture was poured into 300 ml of ice water with great caution keeping the temperature of the solution below 60°. The solid thus obtained was collected by filtration and washed with plenty of water and dried. The solid was recrystallized from benzene to give 0.81 g (2.65 mmoles, 37%) of 26 as a yellowish powder, mp 240-242°; ir (potassium bromide): 3080 (aromatic CH stretching); ¹H nmr (DMSO-d₆): 130° δ 7.26-7.82 (m, 4H, ArH), 7.96 (dd, $J_{8.9} = 7.8$ Hz, $J_{8.10} = 1.2$ Hz, 1H, H-8), 8.67-8.88 (m, 1H, H-1).

Anal. Calcd. for C₁₅H₆ClF₂NS: C, 58.93; H, 1.98; N, 4.58; S, 10.49. Found: C, 58.81; H, 1.80; N, 4.72; S, 10.48.

6-Chloro-2,11-difluoro[1]benzothieno[2,3-c]quinoline (27).

This compound was obtained from benzene in 83% yield from the corresponding lactam 21 in a manner similar to that described for the synthesis of 26, mp 224-225°; ir (potassium bromide): 3065 (aromatic CH stretching); 1 H nmr (DMSO-d₆): 135° δ 7.35-7.88 (m, 3H, ArH), 8.02-8.23 (m, 2H, ArH), 8.58 (ddd, $J_{1,F11} =$ 12.0 Hz, $J_{1.F2} = 9.3$ Hz, $J_{1.3} = 2.7$ Hz, 1H, H-1).

Anal. Calcd. for C₁₅H₆ClF₂NS: C, 58.93; H, 1.98; N, 4.58: S. 10.49. Found: C, 58.82; H, 1.83; N, 4.41; S, 10.63.

6-Chloro-4.10-difluoro[1]benzothieno[2,3-c]quinoline (28).

This compound was obtained from benzene in 84% yield from the corresponding lactam 22 in a manner similar to that described for the synthesis of 26, mp 246-247°; ir (potassium bromide): 3086 (aromatic CH stretching); ¹H nmr (DMSO-d₆): 100° δ 7.42-8.02 (m, 3H, ArH), 8.20 (dd, $J_{8.9} = 9.1$ Hz, $J_{8.F} = 5.3$ Hz, 1H, H-8), 8.49-8.66 (m, 2H, H-1 and H-11).

Anal. Calcd. for C₁₅H₆ClF₂NS: C, 58.93; H, 1.98; N, 4.58; S, 10.49. Found: C, 58.81; H, 1.80; N, 4.72; S, 10.48.

6-Chloro-2,10-difluoro[1]benzothieno[2,3-c]quinoline (29).

This compound was obtained from cyclohexane in 67% yield from lactam 23 in a manner similar to that described for the synthesis of 26, mp 272-273°; ir (potassium bromide): 3068 (aromatic CH stretching); 'H nmr (DMSO-d₆): 150° δ 7.46-7.79 (m, 2H, H-3 and H-9), 8.18 (dd, $J_{3.4} = 8.7 \text{ Hz}$, $J_{F.4} = 5.9 \text{ Hz}$, 1H, H-4), 8.27 (dd, $J_{8.9} = 9.0 \text{ Hz}, J_{8.F} = 5.1 \text{ Hz}, 1\text{H}, \text{H-8}, 8.53 (dd, J_{1.F} = 7.9 \text{ Hz}, J_{1.3})$ = 2.6 Hz, 1H, H-1), 8.65 (dd, $J_{F,11}$ = 7.7 Hz, $J_{9,11}$ = 2.3 Hz, 1H,

Anal. Calcd. for C₁₅H₆ClF₂NS: C, 58.93; H, 1.98; N, 4.58; S, 10.49. Found: C, 59.08; H, 1.84; N, 4.61; S, 10.38.

6-Chloro-4,9-difluoro[1]benzothieno[2,3-c]quinoline (30).

This compound was obtained from benzene as colorless needles in 74% yield from the lactam 24 in a manner similar to that described for the synthesis of 26, mp 278-279°; ir (potassium bromide): 3080 (aromatic CH stretching); ¹H nmr (DMSO-d₆): 150° δ 7.40-7.92 (m, 3H, ArH), 8.09 (dd, $J_{8F} = 9.0$ Hz, $J_{810} = 2.6$ Hz, 1H, H-8), 8.68 (br d, $J_{1,2} = 8.2$ Hz, 1H, H-1), 8.9 (dd, $J_{10,11} =$ 9.1 Hz, $J_{F11} = 5.0$ Hz, 1H, H-11).

Anal. Calcd. for C₁₅H₆ClF₂NS: C, 58.93; H, 1.98; N, 4.58; S, 10.49. Found: C, 59.13; H, 1.78; N, 4.68; S, 10.29.

6-Chloro-2,9-difluoro[1]benzothieno[2,3-c]quinoline (31).

This compound was obtained from benzene in 60% yield from the lactam 25 in manner similar to that described for the synthesis of 26, mp 260-262°; ir (potassium bromide): 3065 (aromatic CH stretching); ¹H nmr (DMSO-d₆): 120° δ 7.39-7.80 (m, 2H, H-3 and H-10), 8.04-8.25 (m, 2H, H-4 and H-8), 8.54 (dd, $J_{1,F} = 10.4$ Hz, $J_{1,3} = 2.4 Hz$, 1H, H-1), 8.91 (dd, $J_{10,11} = 9.1 Hz$, $J_{F,11} = 5.0$ Hz, 1H, H-11).

Anal. Calcd. for C15H6ClF2NS: C, 58.93; H, 1.98; N, 4.58; S, 10.49. Found: C, 58.93; H, 1.74; N, 4.64; S, 10.36.

4,11-Difluoro[1]benzothieno[2,3-c]quinoline (32).

A mixture of 0.72 g (2.36 mmoles) of 26, 0.132 g (2.36 mmoles) of potassium hydroxide, 0.1 g of 10% Pd-C, 80 ml of benzene and 80 ml of methanol was stirred under a hydrogen atmosphere at atmospheric pressure and room temperature until the uptake of hydrogen ceased. The catalyst was removed by filtration and the filtrate was evaporated to dryness in vacuo. The residual solid was recrystallized from cyclohexane to afford 0.50 g (1.84) mmoles, 78%) of 32 as an off-white powder, mp 175-176°; ir (potassium bromide): 3047 (aromatic CH stretching); 'H nmr (DMSO-d₆): 100° δ 7.36-7.88 (m, 4H, ArH), 8.12 (dd, J_{8.9} = 7.6 Hz, $J_{8.10} = 1.2 \text{ Hz}, 1\text{H}, \text{H-8}, 8.69-8.90 (m, 1\text{H}, \text{H-1}), 9.54 (s, 1\text{H}, \text{H-6}).$ Anal. Calcd. for C₁₅H₇F₂NS: C, 66.41; H, 2.60; N, 5.16; S, 11.82.

Found: C, 66.36; H, 2.89; N, 4.83; S, 11.50.

2,11-Difluoro[1]benzothieno[2,3-c]quinoline (33).

This compound was obtained from cyclohexane as colorless crystals in 74% yield from 27 in a manner similar to that described for the synthesis of 32, mp 175-176°; ir (potassium bromide): 3078) aromatic CH stretching); $^1\mathrm{H}$ nmr (deuteriochloroform): δ 7.19-7.71 (m, 3H, ArH), 7.82 (dd, $J_{8,9}=7.8$ Hz, $J_{8,10}=1.2$ Hz, 1H, H-8), 8.26 (dd, $J_{3,4}=9.0$ Hz, $J_{F,4}=6.0$ Hz, 1H, H-4), 8.78 (ddd, $J_{1,F11}=12.2$ Hz, $J_{1,F2}=10.3$ Hz, $J_{1,3}=2.7$ Hz, 1H, H-1), 9.24 (s, 1H, H-6).

Anal. Calcd. for C₁₅H₇F₂NS: C, 66.41; H, 2.60; N, 5.16; S, 11.82. Found: C, 66.19; H, 2.84; N, 5.19; S, 11.66.

4,10-Difluoro[1]benzothieno[2,3-c]quinoline (34).

This compound was obtained from cyclohexane as off-white crystals in 82% yield from **28** in a manner similar to that described for the synthesis of **32**, mp 193-195° dec; ir (potassium bromide): 3055 (aromatic CH stretching); 'H nmr (deuteriochloroform): δ 7.23-7.75 (m, 3H, H-2, H-3 and H-9), 7.91 (dd, $J_{8,9}=8.8$ Hz, $J_{8,F}=5.1$ Hz, 1H, H-8), 8.17-8.36 (m, 2H, H-1 and H-11), 9.25 (s, 1H, H-6).

Anal. Calcd. for C₁₅H₇F₂NS: C, 66.41; H, 2.60; N, 5.16; S, 11.82. Found: C, 66.19; H, 2.26; N, 5.09; S, 11.73.

2,10-Difluoro[1]benzothieno[2,3-c]quinoline (35).

This compound was obtained from cyclohexane as colorless crystals in 82% yield from 29 in a manner similar to that described for the synthesis of 32, mp 206-208°; ir (potassium bromide): 3068 (aromatic CH stretching); ¹H nmr (deuteriochloroform): δ 7.29-7.64 (m, 2H, H-3 and H-9), 7.99 (dd, $J_{3,4} = 8.8$ Hz, $J_{F,4} = 5.1$ Hz, 1H, H-4), 8.21-8.43 (m, 3H, H-1, H-8 and H-11), 9.27 (s, 1H, H-6).

Anal. Calcd. for C₁₅H₇F₂NS: C, 66.41; H, 2.60; N, 5.16; S, 11.82. Found: C, 66.40; H, 2.71; N, 5.18; S, 11.76.

4,9-Difluoro[1]benzothieno[2,3-c]quinoline (36).

This compound was obtained from cyclohexane as yellowish crystals in 83% yield from 30 in a manner similar to that described for the synthesis of 32, mp 198-200° dec; ir (potassium bromide): 3080 (aromatic CH stretching); ¹H nmr (deuteriochloroform): δ 7.26-7.81 (m, 4H, ArH), 8.53 (dt, $J_{1,2} = 8.3$ Hz, $J_{1,3} = 1.2$ Hz, $J_{1,F} = 1.2$ Hz, 1H, H-1), 8.73 (dd, $J_{10,11} = 9.0$ Hz, $J_{F,11} = 4.8$ Hz, 1H, H-11), 9.33 (s, 1H, H-6).

Anal. Calcd. for C₁₅H₇F₂NS: C, 66.41; H, 2.60; N, 5.16; S, 11.82. Found: C, 66.32; H, 2.80; N, 5.21; S, 11.62.

2,9-Difluoro[1]benzothieno[2,3-c]quinoline (37).

This compound was obtained from cyclohexane as colorless crystals in 89% yield from 31 in a manner similar to that described for the synthesis of 32, mp 192-194°; ir (potassium bromide): 3060 (aromatic CH stretching); ¹H nmr (deuteriochloroform): δ 7.26-7.76 (m, 3H, ArH), 8.19-8.39 (m, 2H, H-1 and H-8), 8.63 (dd, $J_{10,11}=9.2$ Hz, $J_{F,11}=5.0$ Hz, 1H, H-11), 9.23 (s, 1H, H-6).

Anal. Calcd. for C₁₅H₇F₂NS: C, 66.41; H, 2.60; N, 5.16; S, 11.82. Found: C, 66.27; H, 2.81; N, 5.12; S, 11.66.

4,11-Difluoro-5-methyl[1]benzothieno[2,3-c]quinolinium Iodide (38).

A mixture of 0.34 g (1.25 mmoles) of 32, 2 ml of iodomethane and 25 ml of benzene was heated at 100-110° in a sealed reaction vessel for 24 hours. After cooling the precipitate was collected by filtration and then recrystallized from methanol to afford 0.35 g

(0.85 mmole, 68%) of quaternary salt **38** as orange needles, mp 218-220° dec; ir (potassium bromide): 3050 (aromatic CH stretching), 2993 (aliphatic CH stretching), 1378 (CH bending of CH₃); ¹H nmr (DMSO-d₆): 90° δ 4.85 (d, $J_{F,CH_3} = 9.7$ Hz, 3H, CH₃), 7.59-8.21 (m, 4H, ArH), 8.40 (dd, $J_{8,9} = 8.0$ Hz, $J_{8,10} = 1.0$ Hz, 1H, H-8), 9.01-9.21 (m, 1H, H-1), 10.35 (s, 1H, H-6).

Anal. Calcd. for $C_{16}H_{10}F_2INS$: C, 46.50; H, 2.44; N, 3.39; S, 7.76. Found: C, 46.62; H, 2.19; N, 3.44; S, 7.70.

2,11-Difluoro-5-methyl[1]benzothieno[2,3-c]quinolinium Iodide (39).

This compound was obtained from methanol as orange needles in 71% yield in a manner similar to that described for the synthesis of **38**, mp 294-295° dec; ir (potassium bromide): 3037 (aromatic CH stretching), 2980, 2954 (aliphatic CH stretching), 1376 (CH bending CH₃): ¹H nmr (DMSO-d₆): 100° δ 4.77 (s, 3H, CH₃), 7.56-8.28 (m, 3H, H-3, H-9 and H-10), 8.38 (dd, $J_{8,9}=8.1$ Hz, $J_{8,10}=1.0$ Hz, 1H, H-8), 8.76 (dd, $J_{3,4}=9.7$ Hz, $J_{F,4}=5.1$ Hz, 1H, H-4), 8.98 (dt, $J_{1,F2}\simeq J_{1,F11}=10.5$ Hz, $J_{1,3}=2.8$ Hz, 1H, H-1), 10.44 (s, 1H, H-6).

Anal. Calcd. for C₁₆H₁₀F₂INS: C, 46.50; H, 2.44; N, 3.39; S, 7.76. Found: C, 46.65; H, 2.62; N, 3.47; S, 7.61.

4,10-Difluoro-5-methyl[1]benzothieno[2,3-c]quinolinium Iodide (40).

This compound was obtained from methanol as light orange needles in 24% yield in a manner similar to that described for the synthesis of **38**, mp 209-211° dec; ir (potassium bromide): 3044 (aromatic CH stretching), 1363 (CH bending of CH₃); ¹H nmr (DMSO-d₆): δ 4.85 (d, J_{F,CH₃} = 10.0 Hz, 3H, CH₃), 7.81-8.21 (m, 3H, H-2, H-3 and H-9), 8.64 (dd, J_{8,9} = 9.0 Hz, J_{8,F} = 5.1 Hz, 1H, H-8), 9.00-9.25 (m, 2H, H-1 and H-11), 10.41 (s, 1H, H-6).

Anal. Calcd. for $C_{16}H_{10}F_2INS$: C, 46.50; H, 2.44; N, 3.99; S, 7.76. Found: C, 46.65; H, 2.19; N, 3.43; S, 7.66.

2,10-Difluoro-5-methyl[1]benzothieno[2,3-c]quinolinium Iodide (41).

This compound was obtained from methanol as orange crystals in 74% yield in a manner similar to that described for the synthesis of **38**, mp 292-293° dec; ir (potassium bromide): 3060 (aromatic CH stretching), 1380 (CH bending of CH₃); 'H nmr (DMSO-d₆): 150° δ 4.77 (s, 3H, CH₃), 7.69-8.22 (m, 2H, H-3 and H-9), 8.53 (dd, $J_{3,4} = 9.1$ Hz, $J_{F,4} = 5.0$ Hz, 1H, H-4), 8.75 (dd, $J_{8,9} = 9.7$ Hz, $J_{8,F} = 4.9$ Hz, 1H, H-8), 9.00 (br d, 2H, H-1 and H-11), 10.40 (s, 1H, H-6).

Anal. Calcd. for $C_{16}H_{10}F_2INS$: C, 46.50; H, 2.44; N, 3.39; S, 7.76. Found: C, 46.56; H, 2.67; N, 3.40; S, 7.81.

4,9-Difluoro-5-methyl[1]benzothieno[2,3-c]quinolinium Iodide (42).

This compound was obtained from methanol as yellow needles in 55% yield in a manner similar to that described for the synthesis of **38**, mp 217-218° dec; ir (potassium bromide): 3055 (aromatic CH stretching), 1383 (CH bending of CH₃); 'H nmr (DMSOd₆): δ 4.83 (d, $J_{F,CH_3}=9.7\,$ Hz, 3H, CH₃), 7.77 (td, $J_{10,11}=9.5\,$ Hz, $J_{F,10}=9.0\,$ Hz, $J_{8,10}=2.6\,$ Hz, 1H, H-10), 8.06-8.22 (m, 2H, H-2 and H-3), 8.57 (dd, $J_{8,F}=9.0\,$ Hz, $J_{8,10}=2.6\,$ Hz, 1H, H-8), 9.21 (dd, $J_{1,2}=7.7\,$ Hz, $J_{1,3}=3.0\,$ Hz, 1H, H-1), 9.32 (dd, $J_{10,11}=9.5\,$ Hz, $J_{F,11}=5.1\,$ Hz, 1H, H-11), 10.37 (s, 1H, H-6).

Anal. Calcd. for $C_{16}H_{10}F_2INS$: C, 46.50; H, 2.44; N, 3.39; S, 7.76. Found: C, 46.59; H, 2.34; N, 3.43; S, 7.69.

2,9-Difluoro-5-methyl[1]benzothieno[2,3-c]quinolinium Iodide (43).

This compound was obtained from methanol as yellow crystals in 73% yield in a manner similar to that described for the synthesis of **38**, mp > 320°; ir (potassium bromide): 3060 (aromatic CH stretching), 2988 (aliphatic CH stretching), 1368 (CH bending of CH₃); 'H nmr (DMSO-d₆): 120° δ 4.75 (s, 3H, CH₃), 7.70 (ddd, $J_{10,11}=9.2$ Hz, $J_{F,10}=8.7$ Hz, $J_{8,10}=2.6$ Hz, 1H, H-10), 8.15 (ddd, $J_{3,4}=9.7$ Hz, $J_{F,3}=8.0$ Hz, $J_{1,3}=2.6$ Hz, 1H, H-3), 8.38 (dd, $J_{8,F}=9.1$ Hz, $J_{8,10}=2.6$ Hz, 1H, H-8), 8.75 (dd, $J_{3,4}=9.7$ Hz, $J_{F,4}=5.0$ Hz, 1H, H-4), 9.04 (dd, $J_{1,F}=9.7$ Hz, $J_{1,3}=2.6$ Hz, 1H, H-1), 9.31 (dd, $J_{10,11}=9.2$ Hz, $J_{F,11}=4.9$ Hz, 1H, H-11), 10.33 (s, 1H, H-6).

Anal. Calcd. for $C_{10}H_{10}F_2INS$: C, 46.50; H, 2.44; N, 3.39; S, 7.76. Found: C, 46.39; H, 2.56; N, 3.39; S, 7.65.

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